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Initial Energy Science&Technology(Xiamen) Co., Ltd

PRODUCT CATALOGUE

INNOVATIVE BATTERY TESTING SOLUTION PROVIDER

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(IEST 3 Major Business)

-
- ◆ Battery Testing Service

◆ Battery Testing Instruments

Web: https://www.iestbattery.com

INTRODUCTION **>**

Founded in 2018, Initial Energy Science & Technology Co., Ltd(IEST) is a comprehensive provider of advanced testing instruments for batteries(LIBs, SIBs, and SSBs).

IEST is committed to delivering top-tier testing instruments with following testing scope:

- **Anode & Cathode Powders**: Resistivity & Compaction Density;
- **Sperators/Electrolyte**: Tortuosity / McMullin Number / Ionic Conductivity;
- **Anode & Cathode Electrodes**: Resistance, Uniformity;
- **Cells**: In-situ Gassing & Swelling of coin cells, pouch cells, stacked cells, prismatic cells, cylindrical cells;
- **Modules**: Cyclers, CV & EIS testing.

IEST places significant emphasis on the R&D of cutting-edge technologies, and our mission is to enhance our customers' product quality, so as to contribute to the advancement of new energy technologies, and we have supplied over 2,500 instruments to more than 700 clients worldwide.

CATALOGUE

Material Characterization

Single Particle Force Properties Test System Powder Resistivity & Compaction Density Mea Solid Electrolyte Measurement System Battery Slurry Resistance Analyzer Battery Electrode Resistance Analyzer

In-situ Gassing of Cells

In-Situ Gassing Volume Analyzer In-situ Multi-channel Storage Gassing Test System-Square & Cylinder Cell Internal Pressure Testin

In-situ Swelling of Cells

Model Coin-cell Swelling Analyzer In-Situ Rapid Swelling Screening For Silicon-Ba In-situ Swelling Analyzer for Consumer Cells In-situ Swelling Analyzer for Power and Energy Battery Pressure Distribution Film Cylindrical Cell Swelling Volume Analyzer

Electrochemical Characterization

Electrochemical Property Analyzer Battery Consistency Screening System

Single Particle Force Properties Test System

A Model Table

B Product Introduction

Basic Functions

Apply compression to the particle to generate a force-displacement curve, from which the particle's failure point can be identified. This process determines the force at which the particle is crushed or fails.

Background: Crushing strength of particles can be used to evaluate the pressure resistance of the material and guide the rolling process. Materials with higher particle mechanical strength will have better subsequent cycle stability.

Functional Modules

Displacement, pressure, software integrated control; Real-time photography and video recording of particles.

Applicable Samples

Equipment Schematic

Particle cracked after cyclic compression (SEM Image)

Single Particle Compression (Bottom View)

Scan QR code for details

- 1. Cathode: NMC/LCO/LRMs
- 2. Anode : Silicon-based materials, Hard Carbon, Etc.
- 3. Solid Electrolytes

- 1. Comparison of crushing force distribution: A>B.
- 2. Analysis of stress-displacement curves: Sample A exhibits initial micro-cracking followed by complete collapse, while sample B experiences direct structural collapse and fragmentation.
- 3. Comparison of Disintegration States: After fracturing, all three groups disintegrate into fine granular states.

Sample Preparation: Disperse the powder evenly into the anhydrous ethanol solution, and then add it dropwise to the glass slide;

- **Particle Location**: Locate the single particle with the optical microscope;
- **Particle Compression**: Compress the particle at a constant speed;
- **Data Collection**: Collect the force-displacement curves during the compression so as to find the failure point.

2. Application on Anode Materials ⸺Pure Carbon

Particle Compression Property and Powder Compaction

Conclusion: The compressive property of particle C1 is stronger. hence, C1 powder shows a higher compression modulus than that of C2.

C Main Test Steps D Application Cases

3. Application on Anode Materials ⸺ SiC Composites \blacktriangleright

The two NMC materials A1 and A2 are sintered from the same precursor, but the sintering process is different. The particle size D50 is 18 μm.

Conclusion: The compression resistance of A2 is superior to that of A1, and modifying the sintering process can enhance the material's hardness to a certain extent. Single-particle mechanical property characterization methods offer valuable insights for optimizing the sintering process of materials.

2. Application on Cathode Materials⸺NCM811 N

Conclusion: The crushing processes of the two types of particles differ, leading to variations in powder compaction density and charging cycle performance.

After assembling into half-cells following the same procedure, cycling at 3.0~4.3V, 0.5C, and 45, sample B2 exhibits better cycling stability.

1. Application on Cathode Materials⸺NCM811

Conclusion: SC-3 particles exhibit weaker compressive strength, leading to significantly greater maximum and irreversible deformation during compression compared to the other two samples.

Single Particle Testing

Powder Deformation

Crushed

Result Analysis: Using LCO powder as an example, when the compaction density of the modified powder sample is less than 3.87g/cm³ (pressure <75MPa), its conductivity is lower than that of the unmodified powder sample. However, when the compaction density exceeds 3.87g/cm³ (pressure >75MPa), the conductivity of the modified powder begins to surpass that of the unmodified powder, and the conductivity improves significantly as the compaction increases. Note: Its1 prontizes continuous product updates, and our specifications are subject to change without prior notice.

However, when the compaction density exceeds 3.87g/cm³ (pressure >75MPa), the conductivity of the modif

Result analysis:

Test methods: Put a certain amount of powder $(1-2g)$ into the mold and vibrate it, put the mold into the instrument box, set the pressure (≤200MPa) and the holding time, and start testing the thickness and resistance changes of the powder during the compression process.

Conclusion: When testing the conductivity of powder, the compaction density should be close to the actual compaction of the powder in the electrode.

The compaction density of the LCO electrodes is around $3.8 - 4.0$ g/cm³ after calender.

LCO Powders

Test parameters: Stress, pressure, thickness, resistance, resistivity, conductivity, & compaction density.

Calculation formula

Compaction Density(g/cm³): $\boldsymbol{D} = \frac{\boldsymbol{m}}{S \cdot \boldsymbol{L}}$ Resistance(Ohm): $R = \rho \frac{l}{c}$ Conductivity (S/m): $\sigma_e = \frac{1}{g} = \frac{l}{RS}$ Resistivity(Ω^* cm)-PRCD2100: $\rho = k \frac{U}{I}$ (Where k is the compensation coefficient)

C Why Compaction Density instead of Tapped Density?

B Instrument Principle

Powder Resistivity& Compaction Density Measurement System

A Model Table

Scan QR code for details

The resistivity of the lithium-rich material can be reduced effectively by adjusting its surface structure.

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(3) Silicon-based materials

Analysis of the lithium-rich material with different modification methods.

Conclusion

Resistivity: SiC-1< SiC-2< SiC-3 Compaction density: SiC-1> SiC-2> SiC-3

Conclusion

Resistivity: SiC-1> SiO-2> SiO-3> SiO-4 Compaction density: SiC-1> SiO-2> SiO-3> SiO-4

Test Condition: Sintering temperature of SiO Materials: SiO-1< SiO-2<SiO-3<SiO-4

Conclusion 1: The conductivity of B14-13 is superior to that of B14-14. This is primarily due to its lower porosity, which enhances particle contact throughout the compression process, resulting in better conductivity.

Conclusion 2: The compaction density shows minimal difference under high-pressure conditions but varies under low-pressure conditions. This is mainly because samples with a wide particle size distribution have poor flow and rearrangement characteristics, leading to higher porosity and lower compaction density under low pressure.

Application Cases

(1)Cathode material-LMFP

(2)Lithium-rich materials

E Testing Mold

Conclusion: the conductivity of graphite is greater than that of hard carbon, so is its powder compressibility.

Conductivity evaluation of anode & cathode powders for sodium ion batteries : Effectively evaluate the conductivity and compaction properties of Prussian blue and hard carbon under different modification conditions.

(4)Anode & cathode materials for sodium ion battery

(5)Compression properties of carbon materials

(1)Formation of green pellet

The equipment can be used to prepare the green pellet for solid-state batteries.

(2)Electronic conductivity & compaction density

The electronic conductivity of the solid electrolyte under varying pressures can be measured using an external electrochemical impedance spectroscopy (EIS) module.

(5)Ionic conductivity

Testing range: 10MHz~0.1Hz Voltage disturbance: 10mV The electrochemical impedance spectroscopy (EIS) module automatically measures the ionic conductivity of solid electrolytes under varying pressures.

(3)Electrochemical stabilization window

Using the cyclic voltammetry (CV) module, the electrochemical stability window of solid electrolytes can be analyzed under different pressure conditions.

Solid Electrolyte Measurement System

B Application Cases

A Creative Solutions

cycling performance

The charge-discharge (CD) module allows for the analysis of the cycling performance of solid lithium metal batteries under varying pressures and different electrochemical parameters.

This instrument is suitable for testing of various types of solid electrolytes, such as oxides, sulfides and polymers.

Scan QR code for details

Test Methods:Put a certain volume of slurry (~80mL) into the measuring glass, insert a clean electrode pen, start the software, start to test the change of the slurry resistivity at the three pairs of electrodes with time and save it to the file.

Main features:

1. Separate the voltage and current lines, eliminate the influence of inductance on voltage measurement, and improve the accuracy of resistivity detection; 2. The disc electrode with a diameter of 10mm ensures a relatively large contact area with the sample and reduces the test error;

Resistivity (Ω^* cm): $\rho_e = \frac{U}{I} \times \frac{S}{I}$

3. It can monitor the change of resistivity with time at three positions in the vertical direction of the slurry in real time;

Battery Slurry

When the viscosity, concentration and dispersant type of the conductive agent are changed, the resistivity also changes!

In the future, specifications can be formulated for the slurry resistivity of a certain fixed viscosity, and the stability of the slurry process can be monitored!

Graphite Slurry

Application Cases C

Slurry Resistivity Test Principle A

B Specifications

(2)Concentration-viscosity-resistivity correlation

The resistivity of the slurry decreases with the increase of the concentration, and the change of the viscosity is also inversely proportional to the

relationship;

The I-V curve test of these two types of slurries basically conforms to Ohm's law, and the current and voltage have a linear relationship, indicating that the slurries are mainly electronic conductors;

-Middl

Subsequently, a shelving period can be formulated for a certain of slurry according to the change of the resistivity to ensure the uniformity of the slurry!

(3) Slurry settling performance

On the first and fourth day of testing, the resistivity of the upper and middle channels increased, while the resistivity of the lower channel decreased, indicating that after four days of shelving, the slurry shows obvious settlement.

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Scan QR code for details

1:1 dilution

1:2 dilution

LCO Slurry

Test parameters:The battery electrode resistance analyzer (BER series) adopts **the double-plane pressure-controllable disk electrode** to directly measure the overall resistivity of the real electrode, that is, the sum of the coating resistance, the contact resistance between the coating layer and current collector and the current collector resistance.

(1) Material evaluation : correlation between powder conductivity and **Electrode conductivity**

Feature

Conclusion:Electrode conductivity characterization can be used to evaluate the conductivity and dispersion performance of conductive agents!

C Appliction-Material Evaluation

(2)Conductivity evaluation of conductive agents

Constant pressure test Pressure: 25MPa Holding time: 25s Points tested: 15points.

* Coefficient of Variation COV = (Standard Deviation SD / Mean) \times 100%

Battery Electrode Resistance Analyzer

A Model Table

B Testing method and principle

Scan QR code for details

Conclusion:The quality of the first batch of the 10 anode electrodes is not acceptable as its resistivity is outside the normal range.

Conclusion:The resistivity of the NCM electrodes decreases with the increase of Carbon content, and when the content is greater than 5%, the resistivity decreases slightly.

(3)Evaluation of primer coated aluminum foil: bare aluminum foil, carbon coated aluminum foil A, carbon coated aluminum foil B

Conclusion

(3) Separate the resistivity of the A and B coating layers for the double-coating electrode

- 1. Different primer coating processes will change the conductivity of the current collector;
- 2. After coating 1~2μm primer material on the aluminum foil, the conductivity uniformity of the current collector is better;

(1)Uniformity evaluation for the distribution of conductive agent

D Application Case - Process Evaluation

(2)Positive and negative electrodes with different conductive agents

Conclusion

1. When the A side or the B side is facing up alone, the difference in the resistance and uniformity of the electrode sheet is small; 2. The difference between the A side and the B side after folding is mainly due to the difference in the coating on the two sides, so this method can be used to judge the difference in the coating on the AB side;

Conclusion

- 1. The resistance of **cathode** electrodes increases with the number of **cycles**.
- 2. The resistance of **anode** electrodes increases with the **storage time**.

(1)Analysis of electrode resistance during high temperature cycle&storage

E Application Case - Failure Analysis

Constant pressure test Pressure: 25MPa Holding time: 15s Points tested: 5 points for each group.

In-Situ Gassing Volume Analyzer

B Creative solution - in-situ gassing monitor

GVM system

Instrument Principles

Multi-Level Gassing Testing: Material Gassing → Single-Layer Stacked Cell Gassing → Small Pouch Cell Gassing →

- Cylindrical & Prismatic Cell Gassing
- **Multi-Channel Gassing Testing**: Single Channel → 2-Channel → 8-ChannelsTesting
- **Multiple Temperature Settings**:Room Temperature Testing → High and Low Temperature Testing (RT to 85°C with Water Bath Control)
- **Comprehensive Gassing Analysis** : Gassing Volume → Gassing Pressure → Gassing Composition Analysis

GVM丨 20

A Model Table

By combining Newton's law (formula 1) and Archimedes' buoyancy principle (formula 2), specialized sensors are used to measure the real-time mass changes of the cell during the charge & discharge process, and then the cell's volume changes can be further calculated(formula 3 and 4).

Applications

Overcharge Gassing Strage Gassing Storage Gassing Strage Str

Note: IEST prioritizes continuous product updates, and our specifications are subject to change without prior notice.

D Product Features

Maximum Dimensions (Excluding Tabs): 220 \times 180 mm (Custom sizes available upon request)

Scan QR code for details

 Tempreture: 25℃ \odot Rate: 0.5C

Test Conditions

(1)NCM cells with different Ni contents

Conclusion: The gas production increases gradually with the increases of formation temperature, and when formation temperature is around 55℃, the first phase transition reaction peak will be more acute.

In addition, from the differential capacity curve, as the formation temperature increases, the polarization decreases.

Conclusion: Both cathode electrodes and the contents of electrolyte additives affect gas production, while the type of additives mainly affects the potential of gas production.

Note: C: Cathode electrodes E: Electrolyte additives

(1)Formation at different temperatures

(2)Formation with different electrolyte additives

E Applications - Formation Gassing

E Application Case - Overcharge Gassing

21丨 GVM GVM丨 22

Conclusion: The gas production & gas production rate of the cells with additive A(red) are greater than those without the additive, which means this additive enables a more complete **SEI formation** in the cells.

Conclusion

1. The slope of the volume change curve suddenly increases when overcharged to a certain potential, then the surface temperature of the cell increases sharply, and gas generation starts instantly from there; 2. As the nickel content increases, the state of charge (SOC) at the onset of gas generation shifts from 138% to 115%.

(2) Cells with different cathodes and contents of electrolyte additives

(3) Overcharge and overdischarge of LFP batteries

Conclusion

1. As the cell is overcharged or overdischarged, the starting point of gas production can be detected in real time;

2. Gas chromatography analyzes the gas composition under these two working conditions. In addition to the same gas type as the over-discharge cell, a relatively high content of CO and CO2 gas is also detected.

Gassing Testing

Stacked Cell Molds Multi-Channel

High-Precision Pressure Sensors

Test Conditions: ◯ LFP/Graphite Cells ◯ 0.5C CCCV to 5V ◯ 0.5C DC to 0V **Test Conditions:** ◯ 4.2V fullly charged ◯ storage at 85°C for 4h **Conclusion:** Different cathode materials, electrolytes, and storage temperatures all affect the volume change of the cells.

E Applications - Cycling Gassing

(1) Cycle performance of different NCM cells

E Applications - Storage Gassing

(1) Storage performance under 3 different conditions

Test Conditions: © NCM/Gr Pouch Cell © Tempreture: 60°CRate: 0.5C © Voltage: 3-4.2V

E Gassing from silicon-based slurries

F Comprehensive gassing solutions

Conclusion: The volume change of cell B is greater than that of cell A, and the gap of volume change deepens with the increase of cycles, which indicates the irreversible volume swelling increases as well.

Conclusion

1. Pre-magnesium or pre-lithiation treatment of silicon monoxide results in gas generation in the slurry. 2. Lithium compensation additives in the cathode tend to decompose and generate gas during the actual slurry and

25丨 MIG PBP丨 26

Conclusion: As the cycle count increases, the pressure value rises, and after reaching a certain level, it stabilizes for a period of time.

Test conditions: 60°C atmospheric pressure test **Test results:** 115 hours, 0.003MPa fluctuation

Square & Cylinder Cell Internal Pressure Testing System

In-situ Multi-channel Storage Gassing Test System

Feaures

In-situ Storage Gassing testing for **pouch cells** Multi-channel Testing (up to **64 channels**) Automatical Data Recording(volume, voltage and internal resistance) Access to External Cyclers

In-Situ Cell Swelling Solutions

A Complete Solution for Cell Expansion

B Instrument Principle

Model Coin Cell Silicon-Based Anode Consumer Battery

Model coin cell Stacked Cell Pouch Cell Square Cell Short-blade Cell (<700*400*100 mm) Solid-state Batteries

C Specifications

Model Coin-cell Swelling Analyze(MCS Series)

- **In-Situ Rapid Swelling Screening For Silicon-Based Anode(RSS Series)**
- **In-situ Swelling Analyzer for Consumer Battery/Cells (CBS Series)**
- **In-situ Swelling Analyzer for Power and Energy Storage Cells(SWE Series)**
- **Battery Pressure Distribution Film(BPD Series)**

Scan QR code for details

Test Range & Accuracy

-
- Number of channels: 1-4 channels Temperature: -20℃~80℃

 \odot Force: 1kg~10T(Accuracy: 0.3% F.S) \odot Displacement: 0.1mm~100mm Accuracy: $\pm 1\mu$ m

Note: IEST prioritizes continuous product updates, and our specifications are subject to change without prior notice.

Consumer Battery & Power Battery & Energy Storage Battery Battery Pressure Distribution

Measurement System

E Applications-Materials Evaluations

(1) Formation & charge-discharge swelling of cells with different Si/C contents

D Product Features

Test Conditions: Pouch Cell(stacking) 200 mAh (1 cycle) Cathode: NCM811 \odot Anode: 450Si/C (450 mAh/g) 800Si/C (800 mAh/g)

Test Conditions: ⊘NCM/SI/C Coin Cells ⊘ Si/C#D(~14.65um) ⊘ Si/C#C (~11.98um) ⊘ Si/C#B(~4.64um) **Conclusion:** SiC#B shows the minimum swelling volume, and the swelling performance of the 3 anode materials share the

- 1. **Multi-Level Expansion Testing**: Electrodes, Pouch Cell, Prismatic hard shell cell, Short-blade Cell
- 2. **Multi-Channel Expansion Testing**: Single-channel → Dual-channel → Four-channel → Multi-point simultaneous testing
- 3. **Temperature Control**: -20℃—80℃
- 4. **Wide Force Ranges**: 100kg → 300kg → 1000kg → 5000kg → 10000kg

C Equipment Specifications

Conclusion: The higher the **silicon content** in the anode, the greater the swelling is(Max thickness change is around 40%), and the silicon-lithium alloy formed will affect graphite' s **phase transition potential** of lithium intercalation.

(2) Anode: NCM-Si/C cells with different modifications

same trend observed with the SEMs.

Test Conditions: ⊙ NCM523/Gr Prismatic Cells(2400 mAh) © 34cm*46cm*106cm(T*W*L) Tempreture: 0°C/25°C/45°C/60°C

(3) Cycle swelling of cells with different Limetal

E Applications-Process Conditions

(1)Swelling of prismatic cells under different pre-stress

Test Conditions: **⊘NCM523/Gr Prismatic Cells(2400 mAh)** ⊘34cm*46cm*106cm(T*W*L) ⊘Pre-stress: 50N/500N/1000N

Conclusion

1. The proportion of irreversible swelling of the cells can be reduced by increasing the pre-stress. 2. During the charge process, the 2 inflection points of the swelling curve correspond to the 2 peaks of the differential capacity curve, indicating that the swelling of the cell is related to the phase transition of lithium intercalation & deintercalation.

(4) Cycle swelling of cells with different binders **(2) Swelling of prismatic cells under different temperature**)

Conclusion: The **irreversible swelling of the cells increases** in both cases when the temperature increases from 25°C to 60°C, as well as decreases from 25°C to 0°C. However, the causes of such swelling under high-temperature and low-temperature conditions may differ.

Conclusion: The modified lithium metal anode can significantly reduce the volume expansion of the cycle process.

Conclusion: The irreversible swelling of the 4 tested cells is similar, and the main difference lies in the **swelling thickness after one cycle of full charge**, that cells with **binder C** outperformed the others

(1)Lithium plating under different rate

Test method:Charge the cells at different rates and discharge them at the same rate to analyze the differences in their voltage curves and swelling thickness curves.

(2)Lithium plating under different tempreture

Test method:In situ detect the thickness curves of batteries with different temperatures.

Conclusion: The position where the thickness curve at a certain temperature bifurcates compared with the thickness curve under high temperature which is without lithium plating is the temperature window of the lithium plating.

1. The slope of the cell's thickness curve increase with the increase of rate.(c) 2. Lithium plating gets more and more serious with increase of rate.

(4)Swelling of prismatic cells under different pre-stress

Test Conditions

- LFP/Gr Prismatic Cells(100 Ah)
- Pre-stress: 15kg/30kg/60kg

Conclusion

1. The s**welling curve** of the cell corresponds to its **capacity attenuation curve**. Generally, when there is a sudden drop in capacity (the intersection point of the 2 curves), it is either due to gas generation or side reactions.

2. Lithium plating may occur after 115th cycle.

Conclusion

Conclusion

1. The **initial gap** of the cells gradually **decreases** with the increase of pre-stress, and the v**ariation in swelling force** becomes more significant during the charge and discharge process.

2. The **charge polarization** of the cells first decreases and then increases with the increase of pre-stress, indicating that a pre-stress of around **30kg** is beneficial for improving **the rate performance of prismatic cells**.

(3) Swelling of prismatic cells under different cycles **E Non-destructive lithium plating analysis**

Conclusion: The swelling ratio of jelly rolls is greater than that of stacked cells, cause the stress in stacked cells can partially release in all directions, resulting in a smaller overall swelling thickness.

(2) Swelling stiffness VS Compression stiffness under constant pressure

Test Conditions: ○Cell:LCO/GR 2400mAh ○Constant pressure:10/30/50/100/200kg

6

Time(h)

8

Model Table

E Applications-Cell structure

Anode A

Anode B

Anode C

10

12

(1) Multi-layer jelly rolls vs. Single-layer stacked cells

8%

ATHK%
4%

2%

0%

 Ω

 $\overline{2}$

 $\overline{4}$

Stacked cell expansion

Conclusion

1. The expansion stiffness changes regularly with charging and

discharging.

2. The difference between expansion stiffness and compression

stiffness is obvious.

(1) Application

(2)Features

Real-time display of force-time curve Real-time synchronization of charge & discharge data. One click test data export

Visualization of cell pressure distribution (Uniformity)

Multiple Measurement Ranges, Multiple Sensing Points, Multiple Software Features!

Note: IEST prioritizes continuous product updates, and our specifications are subject to change without prior notice.

BPD丨 36

Distribution film + data collector

Charge **Discharge** Stiffness(KN/mm)
8 8 8 8 8 8 8 8 8 8
8 9 9 9 9 9 9 8 9 9 9 Stiffness(KM/mr
일본 용품 용품 및
- 이후 이후 이후 이후 $-1 - 50 - 10$ $\overline{1}$ $- - 100.10$ $-1 - 200 - 10$ $(8, 2, 2, 3, 4)$
 $(8, 2, 2, 2, 3, 4)$
 $(8, 2, 3, 4)$
 $(8, 2, 3, 4)$ $\begin{array}{cccccc}\n\text{Stiff}\n\text{recoStM/mm} \\
\text{in} & \text{in} & \text{in} & \text{in} & \text{in}\n\end{array}$ $\frac{1}{100\%}$

10

Cylindrical Cell Swelling Volume Analyzer

A Product Features

- **D** Optical Imaging + 3D Reconstruction + Real-time Online Monitoring
 D Application Case
- ▶ Non-contact, Non-destructive
- \blacktriangleright High-throughput testing, suitable for mass production

B Model Table

Li-ion
Cells

Real time reconstruction of battery surface morphology and calculation of volume deformation during charge and discharge processes. Combining voltage and current data to detect and predict battery health condition from a higher dimension.

> 21700 Cell parameters:Sample 1-15%SiC ; Sample 2-10%SiC The volume swelling curve during formation shows that as the silicon content increases, the volume swelling during formation process increases, and the peak corresponding to lithium intercalation on the differential capacity curve becomes higher.

High-precision Detection Technology Rotational 3D Reconstruction Technology

dQ/dV Vs AV 0.5 0.4 $0.3 \&$ 0.2×10^{-10} $---S1 dQ/dV$ $---S2 dQ/dV$ $-$ S1 Δ V/V0 $-$ S2 Δ V/V0 2.0 3.0 3.5 2.5 4.0 Voltage (V)

Scan QR code for details

Electrochemical Property Analyzer

A Model Table

B High-Precision Current & Voltage Testing

C CV&EIS + Battery Cycler

ECT&ERT丨 40

The 0.01% testing accuracy can precisely measure the specific capacity of new materials and detect subtle side reactions during the initial stages of battery cycling. This allows for a comprehensive performance evaluation and lifetime prediction of the battery in a short period.

Minimize wiring, handling, and temperature adjustments, streamline operations

Scan QR code for details

41丨 ECT&ERT ECT&ERT _ 42

Comparison with well-known foreign brand B electrochemical workstations

- EIS test results show COV within 2%, ensuring high reproducibility compared to other workstations.
- Better SNR in large cell testing than workstations without current amplifiers.

D IEST Innovative Solutions

F Comparison of EIS results with other electrochemical workstations

E Offers common functions of an electrochemical workstation

Equipped with a 24-bit ADC and 16-bit DAC, achieving high-precision voltage and current control and testing.

Time/s

The ERT series includes common electrochemical workstation functions such as CV, LSV, EIS, CA, and CP.

CV test of 120mAh Pouch cell [Scan speed: 1mV/s]

Eliminates switching time between instruments

Case4:CA·CP Test

Background and significance of battery cell consistency testing before shipment B

Battery Consistency Screening System

A Model Table

C Creative Solution

In new energy vehicles or energy storage power stations, lithium batteries are often used in the form of multiple parallel modules or packs. Therefore, high consistency requirements are placed on the battery cells in the same module or pack. Otherwise, thermal runaway may occur easily due to overcharging/overdischarging of a certain battery cell, leading to many after-sales problems.

Electrochemical impedance spectroscopy (EIS) has good sensitivity and correlation with the SOC, SOH, internal temperature, internal short circuit, etc. of the battery cell. By using fast EIS testing and neural network algorithm modeling, you can effectively screen the consistency of the battery cells and help the cascade utilization of the battery cells.

Electrochemical impedance spectroscopy (EIS) can be used to characterize the resistance of electrochemical processes with different time constants. Introducing EIS testing before battery shipment or after battery delivery, and comparing the impedance differences between different batteries, can (1) screen the consistency of batteries; (2) find abnormal batteries; (3) help analyze the failure mechanism of batteries!

Scan QR code for details

E Application Cases

Case 1: Consistency assessment on OCV stage Case 2: Consistency testing of incoming batteries

D BCS6000 Introduction BCS6000 Introduction BUCKS E *Application Cases - EIS Test of Energy Storage Battery*

Functions & Features

- 1. Battery Consistency Screening;
- 2. Wide cell capacity(**2Ah to 1000Ah**);
- 3. Fast EIS frequency sweep testing(**1500Hz ~ 0.1Hz**);
- 4. OCV, DCIR, CC Charge-Discharge, etc;
- 5. Dynamic fitting screening algorithm for batch screening.

Screening Principle

Appearance

The EIS test frequency range can be adjusted according to the production line progress and process section

EIS test of Prismatic Cells with a Capacity of 50 Ah (at 50% SOC) EIS Frequency: 1500 Hz~0.1 Hz

EIS test of Cylindrical Cells with a Capacity of 30 Ah (at 6.5% SOC) EIS Frequency: 1500 Hz~0.1 Hz

恩 Key Achievment

- 700+ Customers served
- 2500+ Instruments supplied
- \emptyset 5 National Testing Standards drafted
- 100+ LIBs Testing Patents granted
- 230+ Application Articles published
- 20+ papers in leading journals published

Featured Clients

